

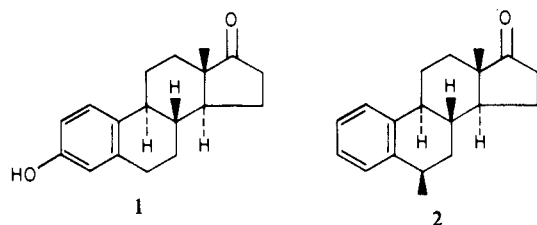
New Stereoselective Synthesis of Steroids

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Synthesis of steroidal systems has been of special interest among synthetic chemists. Some methodologies have been developed for synthesis of steroid frameworks on the basis of intramolecular cyclization of *o*-quinodimethanes, generated by thermolysis of benzocyclobutene precursors^{1a-c} and thermal cheletropic elimination of SO₂ from 1,3-dihydrobenzo[*c*]thiophene 2,2-dioxides.^{1f,g} In the previous paper² we described a simple and versatile method for an efficient generation of *o*-quinodimethanes in which [*o*-[(trimethylsilyl)methyl]benzyl]trimethylammonium iodide was treated with fluoride anion at room temperature. We now wish to report a stereoselective synthesis of estrone (**1**) and 6 β -methyl-estra-1,3,5(10)-trien-17-one (**2**) on the basis of the new methodology.



The key in the construction of the steroidal structures is a selective generation of the silicon-stabilized carbanions from [2-[(trimethylsilyl)methyl]-5-methoxybenzyl]dimethylamine (**3**)³ and α -[*o*-[(trimethylsilyl)methyl]phenyl]ethyl dimethylamine (**4**)⁵ and their alkylations to prepare the requisite precursors **7** and **8** to lead to the polycyclic frameworks.

The generation of the silicon-stabilized benzylic carbanion **6** from **4** was performed by the addition of 1.5 equiv of *n*-butyllithium in THF at 0 °C according to the reported procedure.² On the other hand, an attempt to generate the silicon-stabilized benzylic carbanion **5** from **3** was hampered by kinetically favored lithiation at the carbons ortho to the methoxy substituent on the aromatic ring. Finally, the selective generation of **5** was achieved by treating **3** in THF with *n*-butyllithium in the presence of HMPA [**3** (1 mmol), *n*-BuLi (1.5 mmol), HMPA (1 mL), THF (1.5 mL); between -10 and ~-20 °C; 2 h].

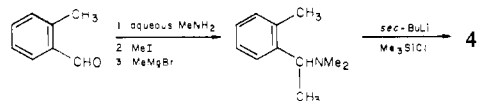
(1) (a) Kametani, T.; Nemoto, H.; Ishikawa, H.; Shirohama, K.; Matsumoto, H.; Fukumoto, K. *J. Am. Chem. Soc.* **1977**, *99*, 3461. (b) Funk, R. L.; Vollhardt, K. P. C. *Ibid.* **1977**, *99*, 5483. (c) *Ibid.* **1980**, *102*, 5253. (d) Oppolzer, W.; Battig, K.; Petrzilka, M. *Helv. Chim. Acta* **1978**, *61*, 1945. (e) For a recent review, see: Oppolzer, W. *Synthesis* **1978**, 793. (f) Nicolaou, K. C.; Barnette, W. E.; Ma, P. *J. Org. Chem.* **1980**, *45*, 1463. (g) Oppolzer, W.; Roberts, D. A.; Bird, T. G. C. *Helv. Chim. Acta* **1979**, *62*, 2017.

(2) Ito, Y.; Nakatsuka, M.; Saegusa, T. *J. Am. Chem. Soc.* **1980**, *102*, 863.

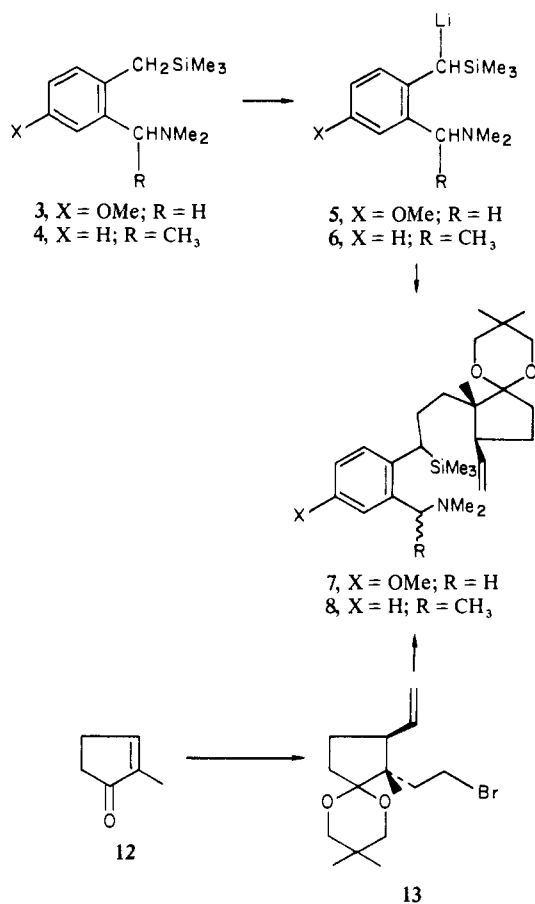
(3) Compound **3** was synthesized in >90% yield by a nickel-catalyzed coupling reaction⁴ of [(trimethylsilyl)methyl]magnesium chloride with (2-chloro-5-methoxybenzyl)dimethylamine which is readily prepared from commercially available 3-methyl-4-chlorophenol. **3**: ¹H NMR (CCl₄ with Me₄Si as an external reference) δ 0.00 (s, 9 H), 2.04 (s, 2 H), 2.10 (s, 6 H), 3.16 (s, 2 H), 3.66 (s, 3 H), 6.3-6.9 (m, 3 H).

(4) Tamao, K.; Sumitani, K.; Kumada, M. *J. Am. Chem. Soc.* **1972**, *94*, 4374.

(5) Compound **4** was synthesized in 95% yield by lithiation (*sec*-BuLi/

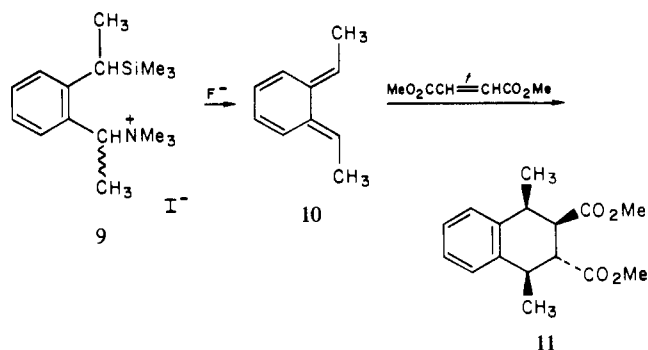


Et₂O) at the *o*-methyl group of [α -(*o*-methylphenyl)ethyl]dimethylamine, which is prepared from *o*-toluylaldehyde, and subsequent treatment with trimethylsilyl chloride. **4**: ¹H NMR (CCl₄ with Me₄Si as an external reference) δ 0.09 (s, 9 H), 1.26 (d, 3 H), 2.13 (s, 6 H + 2 H), 3.33 (q, 1 H), 6.6-7.4 (m, 4 H).



It is to be noted that a 3:2 mixture of diastereoisomers (**9**),⁶ which was prepared by methylation of **6** and subsequent quaternization with methyl iodide, was treated with tetrabutylammonium fluoride in acetonitrile at 50 °C in the presence of dimethyl fumarate to give a single product, which was unambiguously assigned to **11** by NMR spectra,⁷ in a quantitative yield.

This finding indicates that the fluoride anion induced 1,4-elimination of a mixture of diastereoisomers **9** afforded stereoselectively (*E,E*)- α,α' -dimethyl-*o*-quinodimethane (**10**).



This also implies that a variety of symmetrical and unsymmetrical (*E,E*)- α,α' -disubstituted *o*-quinodimethanes may be

(6) **9**: ¹H NMR (CD₃CN, Me₄Si as an external reference) δ -0.27 and -0.09 (s, 9 H), 0.90 and 1.10 (d, 3 H), 1.3-1.6 (br dt, 3 H), 2.0-2.6 (m, 1 H), 2.67 (br s, 9 H), 4.4-5.1 (m, 1 H), 6.6-7.3 (m, 4 H).

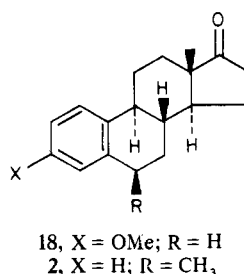
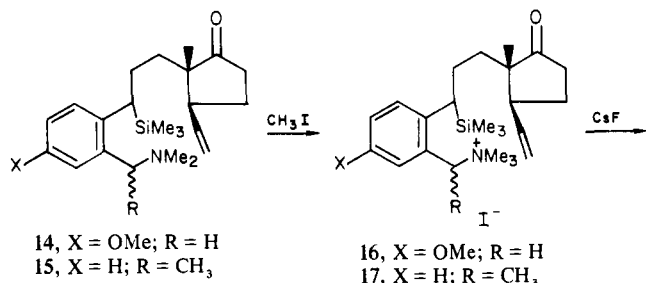
(7) **11**: IR (neat) 1734, 1263, 1196, 1167, 758 cm⁻¹; ¹H NMR (100 MHz, CCl₄) δ 1.10 (d, 3 H, *J*_{H-H_{ax}} = 6.9 Hz), 1.42 (d, 3 H, *J*_{H-H_{ax}} = 6.5 Hz), 2.72 (dd, 1 H, *J*_{H_{ax}-H_{ax}} = 10.0 Hz, *J*_{H_{ax}-H_{ax}} = 11.7 Hz), 2.95 (qd, 1 H, *J*_{H-H_{ax}} = 6.5 Hz, *J*_{H_{ax}-H_{ax}} = 10.0 Hz), 3.12 (dd, 1 H, *J*_{H_{ax}-H_{ax}} = 11.7 Hz, *J*_{H_{ax}-H_{ax}} = 4.8 Hz), 3.29 (qd, 1 H, *J*_{H-H_{ax}} = 6.9 Hz, *J*_{H_{ax}-H_{ax}} = 4.8 Hz), 3.67 (s, 3 H), 3.69 (s, 3 H), 7.9-8.3 (m, 4 H); ¹³C NMR (CDCl₃ with Me₄Si) δ 19.91, 21.03, 35.32, 36.31, 44.99, 46.65, 51.86 (2 C), 126.26, 126.80, 127.33, 128.59, 138.30, 140.01, 173.95, 176.19; TLC on silica gel (8:1 hexane-acetone), *R*_f 0.26.

generated by the present method, since their precursors are readily accessible from *o*-tolualdehyde.⁵

In a parallel line of experiments, cyclopentanone moiety **13** was prepared in an almost stereoisomerically pure form, starting from 2-methyl-2-cyclopentenone (**12**) by modification⁸ of the reported method.^{1c,d,f}

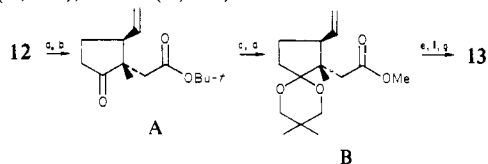
The assembly of the bromide **13** with **3** was carried out by adding **13** to the benzylic carbanion **5** at -75°C to room temperature, which had been generated in situ from **3**, according to the procedure described above. In the assembly of the bromide **13** with **4**, HMPA was added to the carbanion **6**, generated in situ prior to the alkylation. Deketalization of the coupling products **7** and **8** afforded the crucial precursors **14** and **15**¹⁰ as stereoisomeric mixtures [**14**, 2:1;¹¹ **15**, 1:2 (the former is less polar)] in 94% and 95% yields, respectively, after preparative TLC [**14**: TLC on silica gel, R_f 0.60 (6:4 AcOEt-C₆H₆); **15**: TLC on silica gel, R_f 0.59 and 0.38 (6:4 AcOEt-C₆H₆)].

The final intramolecular cyclization of **14** and **15** via the corresponding *o*-quinodimethanes was exemplified with the estrone synthesis. To an acetonitrile solution (20 mL) of a diastereo-



isomeric mixture of **16**, which had been prepared by quaternization of **14** (1.1 mmol) with CH₃I at 0°C , was added a suspension of

(8) The bromide **13**⁹ was prepared stereoselectively in 47% overall yield from 2-methyl-2-cyclopentenone **12** via the keto ester A (>96% stereochemical purity) as shown. **13**: [bp 115–117 $^{\circ}\text{C}$ (0.3 mmHg); mp 53–54 $^{\circ}\text{C}$, recrystallization from hexane]; IR (KBr disk) 1642, 1105, 921 cm^{-1} ; ¹H NMR (CCl₄ with Me₄Si) δ 0.73 (s, 3 H), 0.82 (s, 3 H), 1.19 (s, 3 H), 1.4–2.7 (m, 7 H), 3.0–3.9 (m, 6 H), 4.7–5.9 (m, 3 H).



(a) CH₂=CHMgBr/CuI (3 mol %)/THF. (b) BrCH₂CO₂Bu-*t*/HMPA. (c) MeOH/dry HCl. (d) HOCH₂C(CH₃)₂CH₂OH/CH(OCH₃)₃/*p*-TsOH. (e) LiAlH₄. (f) TsCl/Py. (g) LiBr/DMF.

(9) All attempts to prepare the ethylene glycol ketal corresponding to the bromide **13** from its precursor tosylate and alcohol by conventional methods provided the desired bromide in low yields with several products.

(10) **14**: IR (neat) 1738, 1639, 1246, 862, 838 cm^{-1} ; ¹H NMR (CD₃CN with Me₄Si as an external reference) δ -0.25 (s, 9 H), 0.52 and 0.55 (s, 3 H), 1.1–2.6 (m, 10 H), 2.02 (s, 6 H), 3.08 (s, 2 H), 3.50 (s, 3 H), 4.5–5.0 (m, 2 H), 5.1–5.9 (m, 1 H), 6.3–6.8 (m, 3 H). **15**: IR (neat) (two stereoisomers show almost same spectrum) 1734, 1634, 1242, 856, 831 cm^{-1} ; ¹H NMR (CD₃CN with Me₄Si as an external reference) (less polar isomer) δ -0.26 (s, 9 H), 0.50 (s, 3 H), 0.96 (d, 3 H), 1.2–2.6 (m, 10 H), 1.85 (s, 6 H), 3.20 (q, 1 H), 4.5–4.9 (m, 2 H), 5.0–5.8 (m, 1 H), 6.6–7.2 (m, 4 H); (polar isomer) δ -0.24 (s, 9 H), 0.56 (s, 3 H), 1.07 (d, 3 H), 0.9–2.7 (m, 10 H), 1.99 (s, 6 H), 3.44 (q, 1 H), 4.4–5.0 (m, 2 H), 5.1–5.8 (m, 1 H), 6.6–7.3 (m, 4 H).

(11) The ratio of the stereoisomers was determined by NMR spectrum.

CsF (2.2 mmol) in 10 mL of acetonitrile at once at reflux; the mixture was heated for 1.5 h. The reaction mixture was evaporated in vacuo, triturated with CH₂Cl₂, and filtered to remove the insoluble materials. The filtrate was evaporated and chromatographed on silica gel with chloroform solvent (TLC, R_f 0.45) to afford estrone methyl ether (**18**)¹² in 86% yield based on **14**, which was identified by comparison of its spectral data with those of the authentic sample. The similar treatment of **15** gave rise to 6 β -methylstra-1,3,5(10)-trien-17-one (**2**)¹⁴ in 95% yield. It was identified by comparison of the spectral data with those of the authentic sample, which was independently prepared via hydrogenation on Pd/C of 6-methylstra-1,3,5(10),6-tetraen-17-ol, starting with estradiol 17-monoacetate.^{15,16}

The simplicity and versatility of the present methodology for the generation of *o*-quinodimethane intermediate have been demonstrated by the stereoselective syntheses of estrone and 6 β -methylstra-1,3,5(10)-trien-17-one. Furthermore, the generation of *o*-quinodimethanes with appropriate substituents on the aromatic nucleus and the α and α' carbons is expected in our methodology, which makes derivatization of steroidal skeleton possible.¹⁷ Further extensions of the methodology are now in progress in our laboratory.

Acknowledgment. We are grateful to Shin-etsu Chemical Industry Co., Ltd., for the generous gift of trimethylchlorosilane. We also thank Teikoku Horm. Manufacturing Company for providing estradiol 17-monoacetate, which was used to prepare the authentic sample of 6 β -methylstra-1,3,5(10)-trien-17-one.

(12) Estrone methyl ether thus obtained contained ca. 7–8% of C(9) epimer, but recrystallization from AcOEt afforded pure estrone methyl ether, mp 183.5–185 $^{\circ}\text{C}$ (lit.¹³ 183.2–184 $^{\circ}\text{C}$).

(13) Johnson, W. S.; Banerjee, D. K.; Schneider, W. P.; Gutsche, C. D.; Shelberg, W. E.; Chinn, L. J. *J. Am. Chem. Soc.* **1952**, *74*, 2832.

(14) **2**: mp 89.5–91.5 $^{\circ}\text{C}$, recrystallization from AcOEt; IR (neat) 1738, 758 cm^{-1} ; ¹H NMR (CDCl₃ with Me₄Si) δ 0.84 (s, 3 H), 1.25 (d, 3 H), 1.1–3.2 (m, 17 H), 6.9–7.3 (m, 4 H). ¹³C NMR (CDCl₃ with Me₄Si) δ 13.60, 21.32, 24.29, 25.15, 31.35, 31.71, 33.06, 33.55, 35.48, 44.29, 47.75, 50.13, 124.62, 125.65 (2 C), 128.44, 139.27, 141.52, 220.05.

(15) Japanese Patent Publication 4071, 1963; Teikoku Horm. Manufacturing Company.

(16) (a) Douglas, G. H.; Buzby, G. C., Jr.; Walk, C. R.; Smith, H. *Tetrahedron* **1966**, *22*, 1019. (b) Velarde, E.; Iriarte, J.; Ringold, H. J.; Djerassi, C. *J. Org. Chem.* **1959**, *24*, 311.

(17) We have been informed through a private communication from Professor P. Magnus of The Ohio State University that he has developed a synthetic route to 11 α -hydroxyestrone [*J. Am. Chem. Soc.* **1980**, *102*, 6885] on the basis of our methodology for generation of *o*-quinodimethane intermediates.

Directional Preferences of Nonbonded Atomic Contacts with Divalent Sulfur in Terms of Its Orbital Orientations. 2. S...S Interactions and Nonspherical Shape of Sulfur in Crystals

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Our earlier studies¹ of nonbonded atomic contacts with divalent sulfur in crystals revealed that electrophiles tend to approach sulfur roughly 20° from the perpendicular to the plane through atoms Y–S–Z, whereas nucleophiles tend to approach approximately along the extension of one of the covalent bonds to S, indicating

(1) Rosenfield, R. E., Jr.; Parthasarathy, R. "Program and Abstracts, American Crystallographic Association"; American Crystallographic Association: 1975; Series 2, Vol. 3, p 28. Rosenfield, R. E., Jr.; Parthasarathy, R.; Dunitz, J. D. *J. Am. Chem. Soc.* **1978**, *99*, 4860–4862.